

# Fe-P-Y (Iron-Phosphorus-Yttrium)

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A partial isothermal section for this system at 800 °C was determined recently by [2002Ori]. The section depicts two ternary compounds.

## Binary Systems

A partial phase diagram is known for the Fe-P system [1982Kub]. The intermediate compound  $\text{Fe}_3\text{P}$  forms through a peritectic reaction at 1166 °C between liquid and  $\text{Fe}_2\text{P}$ .  $\text{Fe}_2\text{P}$  forms congruently at 1370 °C.  $\text{Fe}_3\text{P}$  is body centered tetragonal with the  $\text{Ni}_3\text{P}$ -type structure.  $\text{Fe}_2\text{P}$  has the hexagonal C22 structure. The other intermediate phases at higher P contents are FeP (orthorhombic MnP type),  $\text{FeP}_2$  [orthorhombic  $\text{FeS}_2$  (marcasite) type], and  $\text{FeP}_4$  (monoclinic). The Fe-Y phase diagram [1992Zha] depicts four intermediate phases with limited or no homogeneity ranges:  $\text{Y}_2\text{Fe}_{17}$ ,  $\text{Y}_6\text{Fe}_{23}$ ,  $\text{YFe}_3$ , and  $\text{YFe}_2$ . See [1992Zha] for crystal structure data. The Y-P phase diagram is not known. The compound YP has the cubic NaCl type structure.

## Ternary Isothermal Section

With starting materials of 99.95 Fe, 99.98 P, and 99.5 Y (all at.%), Orishchin et al. [2002Ori] melted 17 ternary alloy compositions in an arc furnace under Ar atm. The alloys were finally annealed at 800 °C for 500 h and quenched in water. The phase equilibria were studied mainly by x-ray

powder diffraction, supplemented by metallography and electron probe microanalysis. The isothermal section at 800 °C determined by [2002Ori] up to 50 at.% P is redrawn in Fig. 1 to agree with the accepted binary data. Two ternary compounds are stable at this temperature.  $\text{Y}_2\text{Fe}_{12}\text{P}_7$  (denoted  $\tau_1$  here) has the hexagonal  $\text{Zr}_2\text{Fe}_{12}\text{P}_7$  type of structure. The crystal structure of  $\text{YFe}_5\text{P}_2$  (the experimentally-determined composition is  $\text{YFe}_{4.8}\text{P}_{1.8}$ ), denoted  $\tau_2$  in Fig. 1, was not resolved. The ternary compound  $\text{YFe}_5\text{P}_3$  [1984Jei] was not found at this temperature. The compound YP forms tie lines with the two ternary compounds as well as with all the binary compounds (except  $\text{Fe}_3\text{P}$ ). The solubility of the third component in the binary phases is negligible.  $\text{Y}_2\text{Fe}_{17}$ , which has two crystal modifications, was found to have the hexagonal  $\text{Th}_2\text{Ni}_{17}$  type structure at 800 °C [2002Ori].

## References

- 1982Kub:** O. Kubaschewski: "Iron-Phosphorus" in *Iron – Binary Phase Diagrams*, Springer-Verlag, Berlin, 1982, pp. 84-86.
- 1984Jei:** W. Jeitschko, U. Meisen, and U.D. Scholtz: "Ternary Lanthanoid Iron Phosphides With  $\text{YCo}_5\text{P}_3$  and  $\text{Zr}_2\text{Fe}_{12}\text{P}_7$ -type Structures," *J. Solid State Chem.*, 1984, 55, pp. 331-36.
- 1992Zha:** W. Zhang, G. Liu, and K. Han: "The Fe-Y (Iron-Yttrium) System," *J. Phase Equilibria*, 1992, 13(3), pp. 304-08.
- 2002Ori:** S.V. Orishchin, O.V. Zhak, S.L. Budnik, and Yu.B. Kuzma: "The Y-Fe-P System," *Zhur. Neorg. Khim.*, 2002, 47(9), pp. 1541-44 (in Russian); TR: *Russ. J. Inorg. Chem.*, 2002, 47(9), pp. 1411-14.

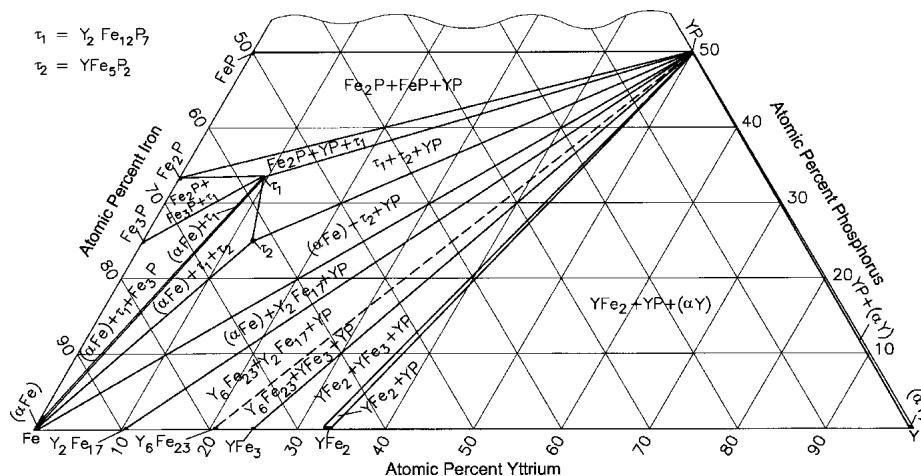


Fig. 1 Fe-P-Y isothermal section at 800 °C [2002Ori]. Narrow two-phase regions around tie-triangles are omitted